IN THE CLAIMS

The following claims listing replaces all prior claims listings:

1 - 4. (Canceled)

5. (Currently amended) A positive electrode active material containing comprising grains, said grains comprising a compound represented by the general of formula Li_x(Fe_yM_{1-y})PO₄, wherein:

 $0.9 \le x \le 1.1$ and $0 < y \le 1$;

M contains at least one 3d transition metal; and

said grains have a grain size not larger than 10 µm., with M containing a 3d transition metal, wherein, in a spectrum for said Li_x(Fe_yM_{1-y})PO₄ obtained by the Moessbauer spectroscopic method, A/B is less than 0.3, where A is the area strength of a spectrum obtained by the Moessbauer spectroscopic method of not less than 0.1 mm/sec and not larger than 0.7 mm/sec and B is the area strength of a spectrum obtained by the Moessbauer spectroscopic method not less than 0.8 mm/sec and not larger than 1.5 mm/sec.

6. (Currently amended) The positive electrode active material according to claim 5 wherein said is Li_x(Fe_xM_{1-x})PO₄ is LiFePO₄. M is one or more of Mn, Co and Ni.

7 - 10. (Canceled)

11. (Currently amended) A non-aqueous electrolyte secondary battery comprising a positive electrode, having a said positive electrode comprising a positive electrode active material, said positive electrode comprising grains, said

grains comprising a compound of formula $\text{Li}_x(\text{Fe}_y\text{M}_{1-y})\text{PO}_4$, where $\text{in}_x(\text{Fe}_y\text{M}_{1-y})\text{PO}_4$, where $0.9 \le x \le 1.1$ and $0 < y \le 1$, with M containing a 3d transition metal, a negative electrode having a negative electrode active material, said positive electrode active material and the negative electrode active material being capable of reversibly doping/undoping lithium, and a non-aqueous electrolyte, wherein, in a spectrum for said $\text{Li}_x(\text{Fe}_y\text{M}_{1-y})\text{PO}_4$ obtained by the Moessbauer spectroscopic method, A/B, A/B is less than 0.3, where A is the area strength of a spectrum obtained by the Moessbauer spectroscopic method not less than 0.1 mm/sec and not larger than 0.7 mm/sec and B is the area strength of a spectrum obtained by the Moessbauer spectroscopic method not less than 0.1 mm/sec and not larger than 0.7 mm/sec and B is the area strength of a spectrum obtained by the Moessbauer spectroscopic method not less than 0.8 mm/sec and not larger than 1.5 mm/sec.

 $0.9 \le x \le 1.1 \text{ and } 0 < y \le 1$;

M contains at least one 3d transition metal; and said grains have a grain size not larger than 10 μm.

- 12. (Currently amended) The non-aqueous electrolyte secondary battery according to claim 11 wherein said Li_x(Fe_yM_{1-y})PO₄ is LiFePO₄. M is one or more of Mn, Co and Ni.
- 13. (Original) A method for producing a positive electrode active material comprising: a mixing step of mixing a starting materials for synthesis of a compound represented by the general formula $\text{Li}_x\text{M}_y\text{PO}_4$, where $0 < x \le 2$ and $0.8 \le y \le 1.2$, with M containing a 3d transition metal; and a sintering step of sintering and reacting said precursor obtained in said mixing step; wherein, in said sintering step, said precursor is sintered at a temperature not lower than 400 °C and not higher than 700 °C.
- 14. (Original) The method for producing a positive electrode active material according to claim 13 wherein, in said sintering step, said precursor is sintered at a temperature not lower than 400 °C and not higher than 600 °C.

15. (Original) The method for producing a positive electrode active material according to claim 13 wherein said $\text{Li}_x\text{M}_y\text{PO}_4$ is LiFePO_4 .